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Controlled Atmosphere Storage of 'Redcoat' Strawberry Fruit

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Abstract. Strawberries (Fragaria x ananassa Duch.) cv. Redcoat were stored at several temperatures and for various intervals in controlled atmospheres (CA) containing 0% to 18% CO, and 15% to 21% 0,. Bioyield point forces recorded on the CA-stored fresh fruit indicated that the addition of CO, to the storage environment enhanced fruit firmness. Fruit kept under 15% CO, for 18 hours was 48% firmer than untreated samples were initially. Response to increasing CO, concentrations was linear. There was no response to changing 0, concentrations. Maximum enhancement of firmness was achieved at a fruit temperature of 0C; there was essentially no enhancement at 21C. In some instances, there was a moderate firmness enhancement as time in storage increased. Carbon dioxide acted to reduce the quantity of fruit lost due to rot. Fruit that was soft and bruised after harvest became drier and firmer in a CO,-enriched environment.

The 'Redcoat' strawberry, widely grown in Ontario, is soft and does not retain good quality during marketing. A means of retaining fruit quality and enhancing firmness would make the berries more acceptable to retailers and consumers. Harris and Harvey (1973) and Li and Kader (1989) reported that CO₂helps prevent softening in fresh fruit in storage. Investigations conducted in Poland (Lange et al., 1978; Ptocharski, 1982) and preliminary investigations conducted in this laboratory (R. B.S., unpublished data) indicated that CO₂ actually will induce firming in strawberries, causing them to become firmer than they were at harvest.

The objectives of this investigation were to 1) test whether storage in CO₂= enriched atmospheres using various temperatures and exposure intervals enhances firmness of 'Redcoat' strawberry fruit and 2) determine whether any firmness enhancement is lost once the fruit is removed from the CO₂ environment. Other quality characteristics were observed to determine whether the CO₂ exposure caused any deleterious changes to the fruit.

Materials and Methods

'Redcoat' strawberries were harvested in the early morning from a commercial planting and transported to the laboratory within 1 h. Containers of 400 to 450 g of fruit were randomized among the various treatments and forced-air cooled to treatment temperatures. Samples were kept for 1 h at the desired temperature to ensure equilibration and then sealed for storage in polyethylene units equipped with inlet-outlet ports (Smith and Reyes, 1988). Atmospheres were monitored with a Nova 305 portable CO₂/O₂ analyzer (Nova Analytical Systems, Hamilton, Ont., Canada) and adjusted with compressed N₂, O₂, and CO₂. Zero percent CO₂was maintained with hydrated lime. In each trial, the treatments were replicated three times.

CO₂ and O₂ concentrations (Trial 1). Storage atmospheres contained O%, 9%, 12%, 15%, or 18% CO₂ in all combinations with 15%, 18%, and 21% O₂. Storage was at OC. Eight sub-

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samples from each replicate were stored under each atmospheric condition; four were removed from each unit after 18 h of exposure to the controlled atmosphere (CA) and four were removed after 110 h. Two subsamples from each atmosphere/storage time treatment combination were removed from the cold room just after removal from CA, and two were held at 0C (post-CA storage) for 48 h and then removed. One of the samples was evaluated immediately upon removal from cold storage; the second was kept for 24 h at 20C and then evaluated (evaluation time).

The fruit from each treatment was first graded into categories of physical condition: good, soft, and decayed. A berry was considered soft if ≥25% of the surface area exhibited signs of excessive bruising (soft and wet). Since fruit temperature affects firmness (Blanpied et al., 1978; Bourne, 1982), a 15-berry sample of the good, unbruised fruit was kept in a IC room for 1 h for temperature equilibration before conducting firmness determinations. Firmness, recorded on a 15-berry sample as the bioyield point force—the force required to initiate shearing (Voisey et al., 1972) (Ottawa Texture Measuring System, Canner's Machinery, Simcoe, Ont.) —was measured on the fruit shoulder through the skin using a 6.35-mm flat tip probe.

Samples of 100 g of fruit from each treatment and replicate for berries kept 18 h in CA and 18 h plus 24 h at room temperature were frozen (– 30C) for subsequent textural studies. Portions consisting of half berries of numerous fruits from one treatment were combined to give a broad-based representative sample. These samples were microwave-thawed (two defrost cycles of 1 min each) and evaluated using a Kramer shear cell adapted to the Ottawa Texture Measuring System (Voisey et al., 1972). Firmness was measured as a function of maximum force applied to shear the thawed sample.

The data from this trial were tested by means of a split plot analysis of variance (ANOVA) with the Statistical Analysis System (SAS) program package (SAS Institute, Cary, N.C.). The magnitude and precision of the experiment were such that the error mean square was very small. As a result, there were many interactions representing responses that were very small in magnitude (<5% change) and thus of little physiological importance and providing little information on elucidation of objectives. Therefore, the data were reanalyzed as if two separate experiments had been conducted. Only those factors significant relative to objective 1 (CO₂ effects on berry firmness) were considered. Factors in the analysis included CA storage time, CO₂, and O₂ for 0 h post-CA storage and 0 h evaluation time.

Those factors having direct bearing on the second objective (firmness changes after removal from CA) were post-CA storage time and evaluation time. Since time in CA storage could have an effect, this factor was also included in the analysis. Data for this analysis were selected to represent CA storage conditions of 15% CO₂ and 18% O₂, the atmosphere used in Trial 3.

CO₂concentration and temperature (Trial 2). The samples were cooled to a pulp temperature of 21, 14, 7, or 0.5C and then placed in treatment rooms at 21, 14, 7, or 0C for 1 h. The samples were then sealed in ployethylene storage units. We used 0%, 6%, 12%, or 18% CO₂at each storage temperature. Oxygen was maintained at 15% for all levels of CO₂. Samples for analysis were removed after exposure to CO₂ for 4, 9, 21, or 30 h.

Bioyieid force measurements (15-berry sample) were recorded with the skin intact and on internal flesh after removal of a slice of skin. Bioyields were determined with skin removed for samples from all treatment combinations, except 9 h. To allow for temperature equilibration for firmness measurements, all fruit was placed in a 0C cold room for 2 h after removal from the respective cold rooms. A 100-g sample composed of portions of numerous berries that had been decapped, washed, and blotted dry, was fast-frozen (– 30C) and subsequently analyzed for ammonium oxalate-soluble and water-soluble pectins by methods of the International Federation of Fruit Juice Producers (1964). Sample preparation for pectin analysis consisted of intermittent microwave heating (with occasional stirring for uniform heating) of the sample to 80C to. deactivate enzymes, cooling to 20C, addition of water to correct for moisture loss, and blending for 1 min (Osterizer Galaxie Six).

Degree of precooking (Trial 3). The berries were precooled to 0.5, 2, 4, 6, or 8C and kept for 1 h at 0, 2, 4, 6, or 8C before being placed in polyethylene storage units. The atmosphere in all units was adjusted to 15% CO₂ and 18.5% O₂. The fruit was evaluated after O, 2, 4, 8, 22, or 44 h in the CA atmosphere.

Tests for the bioyield point (15 berries, skin intact) were conducted as noted in Trial 1. Firmness (Voisey et al., 1972) (the prebioyield point slope) was recorded from the force-distance curve for bioyield points. Also, the energy (Joules) required to shear 200-g samples of treated fruit, frozen to -30C and then thawed, was determined using a Kramer shear cell adapted to the Ottawa Texture Measuring System (Voisey et al., 1972). The samples had been thawed in a microwave oven (defrost cycle for 3 rein).

Data were tested with a split plot ANOVA with the SAS program package. Where applicable, regression analyses were conducted or means were compared using least significant difference. When statistical analysis indicated that a common intercept was present, the regressions were forced through that intercept. The quadratic was fitted where a curvilinear response was indicated. When regressions were not significantly different, a common regression was fitted. Interactions and higher order effects that were not significant were incorporated into the error term. Partial R^2 were calculated from the ratio (SSm -SSr) \div (SSm + SSe - SSr) (where SSm = model sum of squares, SSr = rep sum of squares, and SSe = error sum of squares). In Trial 1, initial samples were not evaluated for all possible levels of each factor, only representative samples for each replicate were tested. Thus, these means could not be included as a factor in the ANOVA. Comparison of initial means to the treatment means of stored samples was done by t test; no such comparisons were made among stored sample means.

Results

Fruit exposed to 15% CO₂ for 18 or 110 h were firmer than the initial samples (Table 1). As the concentration of CO₂ in the storage atmosphere increased, the fruit became significantly firmer and quantity of decayed and soft fruit decreased (Table 1). The response to increasing concentrations of CO₂ was linear (Table 1) for firmness, decay, and soft fruit. Based on regression equations, as storage atmosphere CO₂ increased from 0% to 18%, firmness increased by 29%, decay decreased by 49%, and the portion of soft fruit decreased by 42%. Fruit in CA storage for 110 h compared with 18 h was firmer, had more decayed but less soft fruit. The concentration of O₂ had no effect on firmness or quantity of soft or decayed fruit.

When CA storage time was included as a factor with evaluation time and post-CA storage time, CA storage time had no effect on firmness or quantity of decayed fruit (Table 2). After 24 h at 20C compared with 0 h, fruit had become firmer and the quantity of decay had increased during this interval. Post-CA storage time at 0C had no effect on firmness or quantity of soft fruit; however, there was more decay in the fruit kept for 48 h than in fruit kept for 0 h. An interaction between CA storage time and evaluation time (Table 3) indicated that there was less soft fruit in samples evaluated immediately after 110 h of storage than in those evaluated after 24 h at room temperature or in those stored in CA for 18 h.

Table 1. Comparison of 0 h untreated samples (initials) with the average firmness of stored strawberries and effects of CO₂ and O₂ concentration in the storage atmosphere and duration of exposure to controlled atmosphere on firmness (bioyield point force) and quantity of soft and decayed fruit at 0 h post-CA cold storage and evaluation after 0 h at 20C. (Trial 1).

| Factor | Firmness (N) ^z | Decayed fruit (% by wt) | Soft fruit (% by wt) |
|------------------------------------------|---------------------------|----------------------------|-------------------------|
| Initialy | 2.7 | N/A | N/A |
| Mean for stored samples | 4.5 | N/A | N/A |
| CA storage (h) (S)w,x | | | |
| 18 | 3.71 a | 0.79 a | 3.01 a |
| 110 | 3.97 b | 1.41 b | 1.28 b |
| CO ₂ concn ^{w,v} (%) | | | |
| 0 | 3.25 | 1.72 | 3.11 |
| 9 | 3.74 | 1.03 | 2.00 |
| 12 | 3.94 | 1.00 | 1.99 |
| 15 | 4.15 | 0.88 | 1.83 |
| 18 | 4.18 | 0.87 | 1.80 |

²Means presented for firmness are for the original data, ANOVA conducted using log N. Bioyield force = force required to initiate shearing of flesh using a flat-tip probe.

YComparison by t test (P = 0.05) of initial sample firmness to the mean firmness of stored samples, the latter stored under 15% CO_2 plus 18% O_2 for 18 or 110 h and evaluated just after removal from CA or after 48 h at room temperature. No significant difference between means within stored samples.

*Mean separation by ANOVA (P=0.05). Number of observations per mean for S=45, $O_2=30$, and $CO_2=18$; each observation mean of 15 berries.

*Nonsignificant interactions and higher-order effects incorporated into the error term.

 v Partial R^{2} and linear regressions for significant main effects, no interactions present.

| Relationship | | | R ² |
|--------------|---|-----------------------|----------------|
| Firmness (N) | = | $3.26 + 0.055 [CO_2]$ | 0.72 |
| Decayed (%) | = | $1.62 - 0.049 [CO_2]$ | 0.59 |
| Soft (%) | = | $2.95 - 0.074 [CO_2]$ | 0.31 |

Table 2. Effect of CA storage time, post-CA cold storage, and evaluation time on firmness (bioyield point force) and quantity of soft and decayed fruit stored with 15% CO₂ plus 18% O₂ (Trial 1).

| Factor | Firmness (N)z,y | Decayed fruit (% by wt) ^y | Soft fruit (% by wt) ^y |
|------------|-----------------|--------------------------------------|-----------------------------------|
| CA storag | e (h) | | |
| 18 | 4.48 a | 1.69 a | Int.x |
| 110 | 4.55 a | 1.93 a | Int. |
| Evaluation | n (h) | | |
| 0 | 4.38 a | 0.95 a | Int.x |
| 24 | · 4.67 b | 2.68 b | Int. |
| Post-CA s | storage | 4 | |
| at 0C (1 | h) | | |
| 0 ` | 4.40 a | 1.44 a | 1.41 a |
| 48 | 4.63 a | 2.18 b | 1.20 a |

²Means presented for firmness are for the original data, ANOVA conducted using log N.

*Interaction, CA storage time \times evaluation time significant; see Table 3.

Table 3. Interactive effects of time in controlled-atmosphere (CA) storage and evaluation time on percentage of soft fruit (Trial 1).

| CA storage (h) | Evaluation (h) | Soft fruit (% by wt) |
|----------------|----------------|-------------------------|
| 18 | 0 . | 1.65 a |
| | 24 | 1.23 ab |
| 110 | • 0 | 1.62 b |
| • | 24 | 1. 7 2 a |

²Means separated by LSD (0.94) (P = 0.05). Number of observations per mean = 6, each observation an average of 15 berries.

e maximum force required to shear frozen and subsequently thawed samples was not affected by storage factors.

When firmness was measured with the skin intact or removed, fruit in CA storage became firmer as time in CA storage increased (Table 4). The response was linear. For fruit with skin removed there was a linear increase in firmness as concentration of CO₂ increased. However, there was a linear decrease as temperature of fruit increased. With the skin intact, CO₂ concentration and fruit temperature interacted to affect firmness (Table 5). As temperature of the fruit increased, the firming effect of the CO₂ was dimished. The fruit stored under 12% and 18% CO₂ were firmer than fruit stored under 6% CO₂ (equations significantly different from each other), while the fruit under 0% CO₃ was the softest.

There was a linear increase in the quantity of ammonium oxalate- and water-soluble pectins, the former in response to CA storage time and the latter in response to storage temperature (Table 4). Concentration of CO₂ had no effect on either fraction.

In Trial 3, firmness (bioyield point force and prebioyield point slope) was affected by an interaction between CA storage time and temperature of the fruit (Table 6). As temperature of fruit and CA storage time increased there was a curvilinear increase in firmness (bioyield force) at 0 and 2C and a linear response at 4, 6, and 8C. Whenever firmness was measured as a prebioyield slope response on the force-distance curve, a curvilinear response was present for temperatures from 0 to 4C and no significant response at 6 and 8C (zero slope). In this trial storage factors had no effect on decay (Table 7). An interaction between time and temperature affected quantity of soft fruit, but the

Table 4. Effects of CO₂ concentration in the storage atmosphere, time in CA storage, and temperature of fruit during exposure on firmness (bioyield point force) of strawberries (measured with the skin intact and removed) and quantity of ammonium oxalate- and water-soluble pectins in strawberry fruit (Trial 2).

| | | ness (N)z | Pectin fraction (mg galacturonic acid/kg fruit) | | |
|---------------------|----------------------|-------------------|----------------------------------------------------|------------------|--|
| | , | Skin | | Ammonium | |
| Factor | Intact | Removed | Water-solubley | oxalate*-soluble | |
| CO ₂ con | cn (%CO ₂ | ,) ^{v,w} | | | |
| 0 | Int." | 1.74 | 2340 | 3240 | |
| 6 | Int. | 1.84 | 2260 | 3250 | |
| 12 | Int. | 1.96 | 2220 | 3170 | |
| 18 | Int. | 2.10 | 2210 | 3250 | |
| Temp (7 | r; °C)v,w | | | | |
| 0 | Int." | 2.35 | 2080 | 3180 | |
| 7 | Int. | 1.89 | 2290 | 3260 | |
| 14 | Int. | 1.77 | 2360 | 3320 | |
| 21 | Int. | 1.67 | 2310 | 3250 | |
| CA stor | age (S; h) | v,w | , | | |
| 0 | 2.46 | 1.68 | 2270 | 2940 | |
| 4 | 2.51 | 1.76 | 2190 | 3100 | |
| 9 | 2.63 | | | | |
| 21 | 2.81 | 2.02 | 2280 | 3340 | |
| 30 | 2.90 | 1.94 | 2310 | 3250 | |

*Firmness (N) measured at the bioyield point. Means presented are for original data, ANOVA conducted using log N. Nonsignificant higher order effects incorporated into the error term.

 $^{y}CO_{2}$ and CA storage time had no effect on water-soluble pectin by ANOVA (P = 0.05).

 $^{x}CO_{2}$ and temperature had no effect on ammonium oxalate soluble pectin content by ANOVA (P = 0.05).

"Number of observations per mean for $CO_2 = 60$, for T = 60, and for S = 40; each observation an average of 15 berries.

Partial R² and linear regressions for significant main effects.

| Factor | Relationship | | R ² |
|-----------------|-------------------------------------|------------------------|----------------|
| S | Firmness (N) (skin intact) | = 2.47 + 0.05 [Time] | 0.92 |
| CO ₂ | Firmness (N) (skin removed) | $= 1.73 + 0.02 [CO_2]$ | 0.89 |
| S | Firmness (N) (skin removed) | = 1.76 + 0.01 [Time] | |
| T | Firmness (N) (skin removed) | = 2.24 - 0.03 [Temp] | |
| T | Water soluble | = 2141.0 + 11.0 [Temp] | 0.66 |
| S | Ammonium oxalate- soluble pectin | | 0.00 |
| | (mg·kg ⁻¹) | = 3099.7 + 7.03 [Time] | 0.35 |

"Interactions present in this data, $\mathrm{CO_2} \times \mathrm{T}$ interaction data in Table 5.

changes were very minor and of little importance physiologically or practically. (Thus, other than main effect means, statistical data are not presented for quantity soft in Trial 3.) The firmness of fruit stored in CA, then frozen and then thawed, decreased and then increased as CA storage time increased (Table 7).

Discussion

Strawberries stored in CO₂-enriched atmospheres for intervals Up to 5 days were firmer than the fruit was initially. Firmness was influenced by concentration of CO₂, duration of CA storage, and temperature of the fruit. Harris and Harvey (1973) and Li and Kader (1989) reported that CO₂ maintained berry firm-

^yMean separation with factors by ANOVA (P = 0.05). Number of observations per mean = 12, each observation an average of 15 berries.

Table 5. Effects of concentration of CO₂ in the storage atmosphere and temperature of fruit during exposure on firmness (bioyield point force) of strawberries measured with the skin intact.

| | | Firmne | ss (N) ^{z,y} | |
|-----------------|------|--------|-----------------------|------|
| CO ₂ | | Temp | o (°C) | N |
| (%) | 0 | 7 | 14 | · 21 |
| 0 | 2.54 | 2.39 | 2.47 | 2.18 |
| 6 | 3.18 | 2.53 | 2.61 | 2.39 |
| 12 | 3.67 | 2.86 | 2.60 | 2.50 |
| 18 | 3.72 | 2.86 | 2.81 | 2.50 |

*Firmness (N) measured at the bioyield point. Means presented are for the original data, ANOVA conducted using log N. Nonsignificant higher-order effects incorporated into the error term. Number of observations per mean = 15, each observation an average of 15 berries.

^yPartial R^2 and linear regressions of temperature on CO_2 . No significant difference in regressions for 12% and 18% CO_2 ; thus, one equation is used to describe these responses.

| Factor | Relationship | R^2 |
|-----------------------------------------------|------------------------------------------------------------|-------|
| $\overline{\text{CO}_2(0\%) \times \text{T}}$ | Firmness (N) = | 0.68 |
| CO ₂ (6%) × T | 2.56 - 0.015 [Temp] Firmess (N) = $3.04 - 0.034$ [Temp] | |
| CO ₂ (12% and | Firmness $(N) =$ | |
| $18\%) \times T$ | 3.53 - 0.054 [Temp] | |

ness, but Ptocharski (1982) reported that CO_2 actually enhanced firmness. Research reported here indicated that 'Redcoat' berries kept at 0C under 15% CO_2 for 18 h were 48% firmer than samples were initially $[(4.0-2.7) \times 100 \div 2.7 = 48\%]$. The fact that berries did not soften during 3 days after removal from CA storage is of interest physiologically and of importance commercially.

The response to CO₂ decreased as temperatures increased (Table 5; Trial 2). From the regression equations, fruit exposed to 18% CO₂ at 6C would have a firmness of 3.18 N compared with initial sample firmness of 2.46 N. The difference would represent a 26% firmness enhancement, similar to the 20% to 25% firmness enhancement recorded at 6C by Ptocharski (1982) with 'Senga Sengana' berries. In Trial 2, fruit at 0C in 18% CO₂ was 44% firmer than the initial untreated sample. At 0 to 2C, much of the firming appeared to occur within 24 h of exposure to CO₂. At 4 to 8C, berries firmed more slowly, but after 44 h an essentially equal response was attained. At 14C (Table 5) the response was minimal and slow and at 21C essentially nil. Considering that higher temperatures nullify the effect of CO₂ on firming berries, fruit in CA should be at or near 0C to realize the full enhancing potential of the CO₂.

The effect of CA storage time on firmness was inconclusive. In Trial 1, in one instance, there was a statistically significant but minor enhancement (Table 1) and a nonsignificant effect in another (Table 2). This inconsistency in the same trial could be due to interactions or could reflect the fact that time in CA is only marginally significant. In Trial 2, there was a moderate enhancement attributable to time in CA. This increase paralleled an increase in pectin in the ammonium oxalate soluble fraction that possibly could account for the firmness enhancement associated with time. Moisture loss, minimal inside sealed, or indirectly cooled polyethylene storage units could have caused some external drying and firming but should not have had an internal effect where there was enhanced firmness attributable to CA storage time. Thus, the firmness enhancement associated with time was probably not due to moisture loss.

Table 6. Interactive effects of CA (15% CO₂, 18.5% O₂) storage time and temperature of storage on firmness of strawberries as measured by changes in bioyield point force and prebioyield point slope (Trial 3).

| Temp (T) | | 7 | ime (h) in | storage (| S) | |
|----------|-----------|-------------|-------------|------------|---------------|-------------|
| (°C) | 0 | 2 | 4 | 8 | 22 | 44 |
| | Firm | ness, bioy | ield point | force (N)z | ,у | |
| 0 | 2.23 | 2.55 | 2.85 | 3.06 | 3.29 | 3.25 |
| 2 | 2.33 | 2.37 | 2.61 | 3.07 | 3.05 | 3.23 |
| 4 | 2.29 | 2.36 | 2.57 | 2.63 | 3.03 | 3.38 |
| 6 | 2.23 | 2.38 | 2.59 | 2.51 | 2.85 | 3.08 |
| 8 | 2.20 | 2.48 | 2.40 | 2.56 | 2.77 | 3.10 |
| | Firmness, | , prebioyie | eld point s | lope (N·m | $m^{-1})^{x}$ | |
| 0 | 0.200 | 0.220 | 0.250 | 0.270 | 0.273 | 0.250 |
| 2 | 0.213 | 0.233 | 0.240 | 0.267 | 0.260 | 0.263 |
| 4 | 0.200 | 0.207 | 0.240 | 0.250 | 0.250 | 0.260 |
| 6 | 0.197 | 0.217 | 0.230 | 0.217 | 0.240 | 0.247 |
| 8 | 0.197 | 0.217 | 0.227 | 0.237 | 0.223 | 0.257 |

^zMeans presented for firmness are for original data. ANOVA conducted using log N. Nonsignificant higher-order effects incorporated into the error term. Number of observations per mean = 3, each observation an average for 15 berries.

yPartial \bar{R}^2 and regression equations of time (linear and quadratic) on temperature (linear). No significant differences in regressions for 0 and 2C or 4, 6, and 8C; thus, one equation used to describe response for 0 and 2C (quadratic) and one for 4, 6, and 8C (linear).

Factor Relationship
$$R^2$$

 $S(L) \times T(Q) \text{ (0 and 2C)}$ Firmness(N) = 2.40 + 0.82
 $0.064[\text{Temp}] + 0.001[\text{Temp}]^2$
 $S(L) \times T(L) \text{ (4, 6, and 8C)}$ Firmness(N) = 2.40 + 0.34[Temp]

*Partial R^2 and regression equation of time (quadratic) on temperature (linear). No significant differences in regressions for 0, 2, and 4C; thus, one equation to describe response, slopes at 6 and 8C not significantly different from zero; thus, equations not presented.

Factor Relationship
$$R^2$$

 $S(L) \times T(Q)$ (0, 2, and 4C) Firmness(Slope,N·mm⁻¹) = 0.69
0.216 + 0.0043[Temp] - 0.00008[Temp]²

The prebioyield point slope measures resistance to deformation (Voisey et al., 1972). Since firm fruit deforms less than soft fruit when a given force is applied, the firmer the berry the steeper the prebioyield point slope. Prebioyield point slope data support bioyield firmness data trends in that time in CA and temperature of fruit interacted to affect firmness of fruit in CA.

Firmness measurements on the flesh (skin removed) revealed that internal flesh had also firmed as a result of CO₂ exposure. Thus, the firming effect of CO₂ was not a skin toughening phenomenon that might occur with minor dehydration.

Increased resistance to deformation presumably might be caused by changes in turgor or in pectin type or quantity. These factors also may affect bioyield firmness. Turgor is normally a moisture balance variable presumably not affected by CO₂. Ptocharski (1982) showed that there are changes in pectin composition in strawberries subjected to CO₂ and attributed firmness changes to these pectic changes. He also reported an increase in the quantity of pectin in the ammonium oxalate-soluble fraction and a decrease in the water-soluble fraction. Analysis of strawberries in our laboratory did not reveal any changes in the ammonium oxalate- or water-soluble fraction that could be attributed to the CO₂ treatments. There was a significant increase in the am-

Table 7. Effect of CA (15% CO₂, 18.5% O₂) storage time and temperature of fruit in storage on percent decay, percent soft, and the energy required to shear thawed fruit (Trial 3).

| Duration and temp of storage | Decay (%)² | Soft (%)y | Shear energy (J) ^{z,x} |
|------------------------------|---------------|--------------|------------------------------------|
| Time (h) (S)w 0 | 0.30 | 0.39 | 7.17 |
| 2 | 0.10 | 0.67 | 6.41 |
| 4 | 0.09 | 0.80 | 5.30 |
| 8 | 0.12 | 0.73 | 5.20 |
| 22 | 0.05 | 0.51 | 5.64 |
| . 44 | 0.10 | 0.87 | 5.66 |
| Temp (°C) $(T)^w = 0$ | 0.22 | 0.70 | 5.30 |
| 2 | 0.07 | 0.62 | 5.38 |
| 4 | 0.03 | 0.49 | 6.35 |
| . 6 | 0.19 | 0.50 | 6.33 |
| 8 | 0.12 | 1.00 | 6.17 |
| | | | |

 2 CA storage time had no significant effect on quantity of decay; temperature of fruit had no effect on quantity of decay or shear energy. 3 ANOVA indicated presence of a significant S \times T interaction. However, the magnitude of the changes was small and would be of little physiological or commercial significance, thus, S \times T regression data not presented.

*Partial R^2 and regression equations for time (quadratic) for shear energy.

| Factor | Relationship | R^2 | |
|--------|------------------------------------------------------------------------|-------|--|
| S | Shear energy(J) = $6.46 - 0.108$ [Time] + 0.0021 [Time] ² | 0.15 | |

"Number of observations per mean for S=15 and T=18, each observation an average of 15 berries.

monium oxalate soluble pectin fraction that was attributable to the time in CA storage and a significant increase in the watersoluble fraction, attributable to temperature of fruit. The berries used in our investigation possibly were not sufficiently well matched to detect changes. However, considering the complete absence of a relationship between the two pectin fractions and $CO_2(P = 0.90 \text{ for ammonium oxalate- and } 0.38 \text{ for water-}$ soluble fraction) and the significant increase in firmness as CO₂ concentration increased, it appears that with 'Redcoat' other compositional factors may be a factor. When exposed to CO₂, strawberries may become firmer by the same mechanism or biochemical change that induces firming in tart cherries after they have been bruised (Matz, 1962). As with cherries, where the change is more pronounced at IC than at temperatures > 10C, firming in strawberries was more pronounced at 0C than at 14 or 21C.

In Trial 1 there was no indication that storage factors had any effect on the firmness of berries that had been frozen and thawed. Generally, firmer fresh fruit resulted in firmer frozen thawed

fruit. The texture of frozen thawed fruit cannot be improved by treating fresh fruit with CO, before freezing.

The quantity of decayed fruit increased with CA storage time but decreased as concentration of CO₂ in the storage atmosphere increased. This response to CO₂ has been recorded previously (Couey and Wells, 1970; Harris and Harvey, 1973) and has been attributed to CO₂ suppressing fungal growth.

In Trial 1, the quantity of bruised fruit graded unacceptably soft decreased as time in CA storage increased and as CO₂ concentration increased. This phenomenon appeared to be due to a drying and presumably firming of wounded tissue. This drying and firming effect of CO₂ would provide further justification for using CO₂ in storing and shipping strawberries.

There was no indication that the CO₂ caused any undesirable changes in quality of the fruit. Organoleptic data (R.B.S., unpublished) "showed that the CO₂did not adversely affect fruit quality. The positive response of 'Redcoat' fruit to CO₂ and possibly that of other soft-fruited cultivars, justifies the recommendation that such fruit be stored in a CO₂-enriched atmosphere for several hours after harvest or be shipped under CO₂-enriched atmospheres.

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