

High Nutrient Concentrations of Hydroponic Solution Can Improve Growth and Nutrient Uptake of Spinach (*Spinacia oleracea* L.) Grown in Acidic Nutrient Solution

Daniel P. Gillespie, Gio Papio, and Chieri Kubota

Department of Horticulture and Crop Science, The Ohio State University, Columbus, OH 43210

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Abstract. Hydroponic leafy green production offers high productivity and quality of crops but requires good management of pH and electrical conductivity (EC) to optimize the nutrient uptake. Nutrient solution pH is typically managed between 5.5 and 6.5, whereas lowering pH to more acidic range (e.g., <5.0) can potentially mitigate problematic waterborne diseases. Plant response to low pH is species specific and generally involves direct effect of increased hydronium ions and indirect effects of pH-dependent factors, such as low cations availability. To develop a new hydroponic nutrient management strategy, ‘Corvair’ spinach plants were grown under pH 4.0, 4.5, 5.0, and 5.5 of a hydroponic nutrient solution using a deep-water culture system in a growth chamber. Spinach shoot and root mass after 19 to 20 days declined with lowering pH. At the lowest pH of 4.0, plants displayed stunted overall growth and severely inhibited root development. Plant growth and morphology at pH 4.5 or 5.0 were normal but small, suggesting that growth reduction at these pH was likely a result of reduced nutrient uptake. Plant tissue analyses showed decreased N, P, K, Mg, S, Cu, Fe, Mn, and Zn concentration as pH decreased. When the strength of nutrient solution was increased three times at a low pH 4.5 to improve the overall nutrient availability, spinach shoot and root fresh weight with high nutrient concentrations (EC 3.4 dS·m⁻¹) significantly improved but was still lower than those in the control (pH 5.5 and EC 1.4 dS·m⁻¹), respectively. Plant tissue analysis showed that lowering pH to 4.5 significantly reduced tissue concentrations of P, K, Ca, Mg, S, Cu, Mn, and Zn compared with those in the control. Under low pH and increased EC treatment (pH 4.5 and EC 3.4 dS·m⁻¹), all dry leaf nutrient concentrations were similar or higher than those of the control, except Mg and Zn, which showed a lower concentration than the control with a weak significance ($P < 0.06$). This suggests that additional optimization of nutrient formula might further improve the spinach growth at low pH. Together, our results will help to develop a new and low-cost nutrient management methodology to produce leafy greens hydroponically.

Increased demand for prewashed, prepackaged baby leaves of spinach (*Spinacia oleracea*) over the past decade has led to 51% increase in U.S. spinach production from 2012 to 2017 [Correll et al., 2011; U.S. Department of Agriculture National Agricu-

tural Statistical Service (USDA NASS), 2014, 2019]. Although spinach production and demand has increased, the distribution of spinach production in the United States has changed little, as 93% of fresh market spinach comes from four states, California, Arizona, New Jersey, and Texas (USDA NASS, 2019), where spinach is grown in conventional open field production systems.

In recent years, the growing number of greenhouse and indoor hydroponic operations has increased the availability of clean, locally grown leafy greens. Most leafy green hydroponics operations employ liquid-based cultivation systems such as nutrient film technique and deep water culture (DWC), which allow for efficient water and nutrient use and high productivity. However, although many operations have been successful in growing lettuce (*Lactuca sativa*), arugula (*Eruca sativa*), kale (*Brassica oleracea*), sweet basil (*Ocimum*

basilicum), and other leafy greens, most hydroponic operations, anecdotally, avoid growing spinach due to its susceptibility to *Pythium* spp. infection (Mattson, 2018).

Previous research introduced several mitigation strategies of *Pythium* and other oomycete disease control for spinach (e.g., Albright et al., 2007) and other hydroponically grown crops (e.g., Stanghellini, 1996). Recently, we showed that lowering pH of nutrient solution was a potential control measure for oomycete disease introduction using two cultivars of sweet basil (Gillespie, 2019; Gillespie et al., 2020). In these studies, we found that basil plants could uniquely tolerate pH as low as 4.0, whereas the same low pH was shown to lower the incidence of root rot disease caused by *Pythium aphanidermatum*. In fact, numerous studies have shown the negative effect of low pH (pH <5.0) on oomycete sporangia development and zoospore motility (Blaker and MacDonald, 1983; Ho and Hickman, 1967; Kong et al., 2009). Although the conventional pH 5.5 to 6.5 (Savvas and Gruda, 2018) seems to be a range where almost all hydroponically grown crops exhibit normal growth and nutrient uptake, species-specific pH responses of leafy greens grown in liquid culture hydroponic systems is largely unexplored. Our general understanding is that nutrient disorders and thereby growth reduction occur when pH is outside the optimum range (Adams, 2002). Although commonly referenced pH nutrient availability charts indicate that micronutrients, such as Cu, Zn, Mn, and B availability is increased with decreasing pH (Peterson, 1982), we found that nutrient uptake at low pH declined for many micro and macro elements in our previous experiments for sweet basil (Gillespie et al., 2020). When soilless substrates are used instead of liquid-based hydroponics, pH in the nutrient solution interacts with substrates (Dickson and Fisher, 2019), and micronutrient toxicity occurs rather than deficiency. Therefore, evaluation of plant’s pH response must consider the growing systems employed.

Arnon and Johnson (1942) examined seven levels of pH (3.0–9.0) for lettuce and tomato plants grown hydroponically and showed significant reductions in shoot and root fresh weight below pH 5.0 for both species. However, because pH affects nutrient availability and nutrient uptake across plasma membrane, it is difficult to determine whether growth inhibition and nutrient disorders observed at low pH of the nutrient solution are a result of the direct effect of excessive hydronium ion concentration or pH-dependent factors affecting nutrient availability and uptake. Nevertheless, studies suggest that the direct effect of pH seems to be detrimental only at the extreme ends of acidity and alkalinity, and growth reductions and nutrient disorders outside of the conventional pH ranges can typically be attributed to pH-dependent factors (Arnon and Johnson, 1942; Bugbee, 2004; Gillespie et al., 2020; Islam et al., 1980; Mengel et al., 2001; Vlamis, 1953). Additionally, it has been reported that taking certain precautionary measures to account for pH-dependent factors,

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C.K. is the corresponding author. E-mail: kubota.10@osu.edu.

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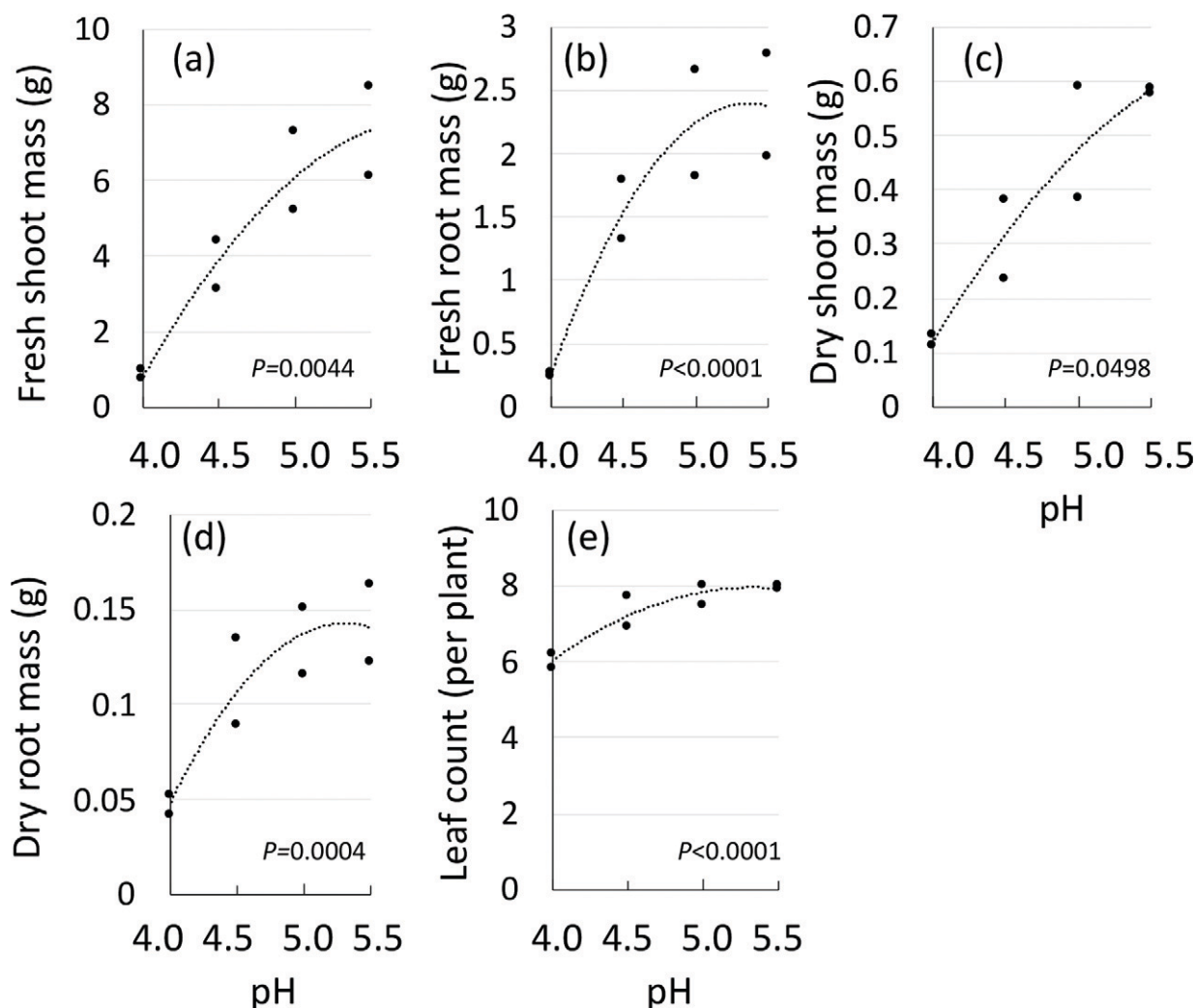


Fig. 1. Spinach plant growth responses to nutrient solution pH 4.0, 4.5, 5.0, or 5.5. Individual charts are (a) fresh shoot mass, (b) fresh root mass, (c) dry shoot mass, (d) dry root mass, and (e) leaf count per plant after 3 weeks. Means of 10 sample plants of two trials are shown. All responses were significant by analysis of variance ($P < 0.15$) and expressed with linear or quadratic regressions: (a) $y = -1.82x^2 + 21.6x - 56.6$; (b) $y = -1.15x^2 + 12.4x - 30.7$; (c) $y = -0.0919x^2 + 1.18x - 3.15$; (d) $y = -0.0553x^2 + 0.587x - 1.41$; (e) $y = -1.10x^2 + 11.7x - 23.2$. Regression P values are shown in each chart.

such as increasing nutrient concentrations in solution, may mitigate pH-dependent factors affecting nutrient availability and uptake. For example, Arnon and Johnson (1942) reported that increasing calcium concentration improved tomato (*Solanum lycopersicum*) and lettuce growth at pH 4.0 and 5.0. Another example by Smith et al. (2004) showed that the higher fertilizer concentrations ameliorated

leaf chlorosis of geranium (*Pelargonium ×hortorum*) plants caused by nutrient deficiencies at pH 7.0 to 7.5.

In the present study, we investigated how pH lower than the conventional range influences spinach plant growth and whether high nutrient concentrations can mitigate growth inhibition and nutrient disorders. Because interactions among uptake of different ions are

complex and increasing hydronium ion concentrations (i.e., low pH) further complicates the relationship, we took a simple approach of increasing the strength of nutrient solution as the first step toward optimization of nutrient formula for low pH applications. Specifically, we examined two strengths of total nutrient concentrations (measured by EC) to grow spinach plants under low pH. Our hypotheses were 1) low pH would reduce nutrient uptake and affect spinach plant growth (Expt. 1) and 2) high nutrient concentrations of hydroponic solution would compensate for low uptake of specific nutrients and improve spinach growth in lower-than-conventional pH (Expt. 2).

Materials and Methods

Plant material, propagation, and water treatment. Spinach ‘Corvair’ seeds (Johnny’s Selected Seeds, Fairfield, ME) were sown in rockwool sheets (AO plugs 200 counts, 2.5-cm height; Grodan, Roermond, The Netherlands) on 4 Nov. 2018 and 29 Dec. 2018 for

Table 1. Aerial environmental parameters recorded in Expts. 1 and 2 (each with two trials).

Trial	Air temp (°C)	Water temp (°C)	VPD ² (kPa)	PPFD (μmol·m ⁻² ·s ⁻¹) ³
Expt. 1				
1	Day: 22.2 ± 0.0 Night: 17.0 ± 0.1	Day: 21.1 ± 0.0 Night: 21.1 ± 0.1	Day: 1.0 ± 0.0 Night: 0.2 ± 0.0	312 ± 1.2
2	Day: 26.2 ± 0.1 Night: 16.6 ± 0.1	Day: 20.8 ± 0.1 Night: 21.1 ± 0.2	Day: 1.2 ± 0.1 Night: 0.2 ± 0.0	324 ± 1.9
Expt. 2				
1	Day: 24.6 ± 0.03 Night: 15.5 ± 0.6	Day: 21.6 ± 0.9 Night: 19.6 ± 2.9	Day: 0.9 ± 0.2 Night: 0.2 ± 0.1	365 ± 5.1
2	Day: 24.7 ± 0.7 Night: 15.3 ± 0.6	Day: 21.6 ± 2.8 Night: 19.4 ± 2.9	Day: 1.1 ± 0.2 Night: 0.2 ± 0.6	346 ± 3.5

²Vapor pressure saturation deficit of air.

³Photosynthetic photon flux density (spatial average measured at 40 locations before and after each trial).

Expt. 1, and 31 July 2019 and 11 Sept. 2019 for Expt. 2. Before seeding, rockwool sheets were placed in white plastic undertrays and hydrated with reverse osmosis water containing $0.5 \text{ mg}\cdot\text{L}^{-1}$ didecyl-dimethyl-ammonium chloride (KleenGrow, Pace Chemicals, Delta, BC, Canada) and allowed to drain. After seeding, trays were placed inside a dark growth chamber set at 20°C air temperature.

For both experiments, except during germination and propagation, municipal water was used and was disinfected with ultraviolet radiation (D4 + Whole Home ultraviolet Water Disinfection System; Viqua, Guelph, ON, Canada). Water used in DWC units was also dechlorinated by the addition of $2.5 \text{ mg}\cdot\text{L}^{-1}$ of sodium thiosulfate. This was done to avoid chlorine phytotoxicity from municipal source water that has been observed in our past experiments. Our water quality analyses typically show low levels of Ca ($<30 \text{ mg}\cdot\text{L}^{-1}$), Cl ($<30 \text{ mg}\cdot\text{L}^{-1}$), $\text{NO}_3\text{-N}$ ($<3 \text{ mg}\cdot\text{L}^{-1}$), Mg ($<8 \text{ mg}\cdot\text{L}^{-1}$), S ($<20 \text{ mg}\cdot\text{L}^{-1}$), K ($<5 \text{ mg}\cdot\text{L}^{-1}$), P ($<0.5 \text{ mg}\cdot\text{L}^{-1}$), Zn ($<0.5 \text{ mg}\cdot\text{L}^{-1}$), Na ($<20 \text{ mg}\cdot\text{L}^{-1}$) and Al ($<0.2 \text{ mg}\cdot\text{L}^{-1}$). Alkalinity is typically $<42.0 \text{ CaCO}_3 \text{ mg}\cdot\text{L}^{-1}$ and approximate EC is 0.2 to $0.3 \text{ dS}\cdot\text{m}^{-1}$.

After radical emergence was observed, seeded trays were moved to the conditions of $25/15^\circ\text{C}$ day/night air temperatures and 12 h/d photoperiod inside a growth chamber (GR96, Conviron, Winnipeg, MB, Canada). Light source in chamber were white fluorescent lamps (Master TL5 54W/840; Philips, Amsterdam, The Netherlands). Seedlings were subirrigated with water as needed until transplanting. The pH of water provided to seedlings before transplant was ≈ 6.4 . When cotyledons were fully expanded, uniform plants with rockwool substrate (15 to 20 mL) were transplanted into DWC units (11 Nov. 2018 and 12 Jan. 2019 for Expt. 1; 9 Aug. 2019 and 25 Sept. 2019 for Expt. 2).

In both experiments, T-type thermocouples were placed at the middle of each side of the chamber for monitoring air temperature at plant canopy level (gauge 36; Omega Inc., Stamford, CT) and nutrient solution temperature (gauge 24; Omega Inc.). Relative humidity was measured with a temperature/humidity probe (HMP60 Humidity and Temperature Probe; Vaisala Corporation, Helsinki, Finland) housed inside an aspirated shield located in the middle of the growth chamber at plant canopy level. Sensors were connected to a datalogger (CR10X dataloggers; Campbell Scientific, Logan, UT) and sensor readings were scanned every 10 s to record averages each 15 min. Vents of the growth chamber were kept open for sufficient outdoor makeup air to provide ambient CO_2 conditions inside the room.

Expt. 1: Effects of nutrient solution pH on spinach plant growth and nutrient uptake. There were four DWC units each with 0.78 m long, 0.51 m wide, and 0.37 m tall black plastic container (Centrex Plastics, LLC Commander 27-Gallon Black Tote; Centrex Plastics, Findlay, OH) and a polystyrene foam raft (Beaver Plastics 72"; Beaver Plastics, Acheson, AB, Canada) cut to match the

size of container. Each DWC unit contained 24 plants in 90 L of nutrient solution made using dechlorinated and ultraviolet radiated water as described previously. The large volume to plant ratio (3.75 L per plant) was to act as a buffer in attempts to minimize pH fluctuations. Nutrient solution was continuously aerated by one air stone connected to a small aquarium air pump.

One-half strength University of Arizona leafy crop nutrient solution recipe (Jensen, unpublished) was used as the basal formula in this experiment. This formula contains ($\text{mg}\cdot\text{L}^{-1}$) $90 \text{ NO}_3\text{-N}$, 25 P , 99 K , 100 Ca , 20 Mg , 1.0 Fe (DTPA-chelated), 0.3 Mn , 0.2 Zn , 0.03 Cu , 0.2 B , and 0.03 Mo . Before transplant, nutrient solution pH was adjusted to set-points 4.0 , 4.5 , 5.0 , and 5.5 using sulfuric acid. After that, pH was monitored at minimum once per day and manually adjusted thereafter as needed by the addition of sulfuric acid or sodium hydroxide to maintain pH within a range of ± 0.25 of target pH. ECs and dissolved oxygen (DO) of each DWC unit were measured at least three times a week. Handheld meters (pH/EC Combo Meter, Bluelab, Tauranga, New Zealand; 407510 DO meter, Extech, Nashua, NH) were used for pH, EC, and DO measurements. EC and pH meters were calibrated weekly. Nutrient solutions were sent to a commercial analytical laboratory (JR Peters, Allentown, PA) for micro- and macronutrients at the end of experiment.

This experiment was replicated over time (Trials 1 and 2). Twenty days (Trial 1) or 19 d (Trial 2) after transplanting, plants were harvested for quantifying plant growth and assessing visible symptoms of nutrient disorders. Fresh/dry shoot and root mass and number of leaves per plant were recorded for 10 randomly sampled plants per pH treatment. Roots were separated from rockwool and carefully dried with paper towels consistently throughout, before fresh weight measurement. Following fresh weight measurements, plant material was dried in a drying oven at 55°C for a minimum of 1 week. Once dry mass was measured, leaf tissue samples of all 10 plants were combined into one sample and sent to the same commercial analytical laboratory to determine nutrient concentrations of leaf tissue.

All data of Trials 1 and 2 were compiled and analyzed as one data set. Location of pH treatments were randomized each replication so that no treatments were located in the same place inside the growth chamber. Plant growth data (fresh/dry mass and leaf number) and leaf nutrient concentration data were evaluated using an analysis of variance (ANOVA) and linear or quadratic regression analysis was applied when ANOVA F -test was significant ($P < 0.15$). All statistical analyses were performed using JMP software (Ver. 14; SAS Institute, Cary, NC).

Expt. 2: Effects of increased nutrient concentrations on spinach growth and nutrient uptake under low pH. Expt. 2 employed four treatments consisting of pH (4.5 or 5.5) and EC (1.4 or $3.4 \text{ dS}\cdot\text{m}^{-1}$) of the nutrient solution using smaller DWC units (36.2 cm tall and

Table 2. Elemental concentrations of nutrient solutions (all values in $\text{mg}\cdot\text{L}^{-1}$) at the end of Expts. 1 and 2. Mean \pm SD ($n = 4$).

Treat. code ^z	Macronutrients ($\text{mg}\cdot\text{L}^{-1}$)										Micronutrients ($\text{mg}\cdot\text{L}^{-1}$)					
	$\text{NO}_3\text{-N}$	P	K	Ca	Mg	S	B	Cl	Cu	Fe	Mn	Mo	Zn			
pH 4.0	87.2 \pm 3.66	23.7 \pm 0.02	104.7 \pm 0.095	144.1 \pm 3.77	26.8 \pm 1.57	59.6 \pm 4.19	0.21 \pm 0.02	88.07 \pm 0.55	0.03 \pm 0.00	1.79 \pm 0.13	0.37 \pm 0.03	0.00 \pm 0.00	0.44 \pm 0.09			
pH 4.5	86.4 \pm 3.65	23.4 \pm 0.35	103.8 \pm 7.13	156.2 \pm 2.18	28.8 \pm 0.20	63.9 \pm 5.42	0.23 \pm 0.01	88.26 \pm 1.61	0.03 \pm 0.01	1.44 \pm 0.09	0.34 \pm 0.03	0.02 \pm 0.01	0.46 \pm 0.14			
pH 5.0	82.8 \pm 3.60	23.0 \pm 0.15	98.0 \pm 8.91	159.5 \pm 1.02	28.4 \pm 0.12	64.7 \pm 1.96	0.22 \pm 0.02	94.17 \pm 5.78	0.03 \pm 0.01	1.27 \pm 0.05	0.33 \pm 0.02	0.02 \pm 0.01	0.45 \pm 0.12			
pH 5.5	81.9 \pm 2.77	21.8 \pm 0.80	86.8 \pm 2.47	149.6 \pm 7.20	26.0 \pm 2.01	59.1 \pm 3.26	0.23 \pm 0.03	91.21 \pm 1.15	0.03 \pm 0.00	1.01 \pm 0.01	0.29 \pm 0.02	0.02 \pm 0.01	0.40 \pm 0.07			
LP/HE	256.7 \pm 5.24	73.1 \pm 1.20	303.7 \pm 6.58	383.0 \pm 5.47	65.9 \pm 1.06	116.6 \pm 0.28	0.50 \pm 0.01	201.31 \pm 5.37	0.04 \pm 0.00	2.94 \pm 0.09	0.85 \pm 0.01	0.08 \pm 0.01	0.73 \pm 0.01			
LP/SE	85.0 \pm 0.82	23.4 \pm 0.11	98.3 \pm 1.04	138.3 \pm 0.50	25.5 \pm 0.04	63.0 \pm 1.16	0.19 \pm 0.01	86.26 \pm 0.88	0.01 \pm 0.00	0.83 \pm 0.01	0.33 \pm 0.01	0.03 \pm 0.01	0.45 \pm 0.06			
SP/HE	263.3 \pm 7.35	73.2 \pm 2.28	304.1 \pm 13.94	387.3 \pm 6.65	66.9 \pm 1.61	109.7 \pm 1.95	0.51 \pm 0.01	208.34 \pm 3.25	0.04 \pm 0.01	2.57 \pm 0.03	0.82 \pm 0.02	0.07 \pm 0.01	0.76 \pm 0.01			
SP/SE (control)	83.1 \pm 4.71	23.1 \pm 0.76	89.7 \pm 11.73	144.3 \pm 0.64	24.7 \pm 0.47	58.0 \pm 0.97	0.19 \pm 0.00	90.25 \pm 3.58	0.01 \pm 0.01	0.67 \pm 0.17	0.28 \pm 0.01	0.03 \pm 0.00	0.50 \pm 0.11			

^zTreatments in Expt. 2 are combinations of LP (low pH 4.5), SP (standard pH 5.5), HE (high electrical conductivity 3.4), and SE (standard electrical conductivity 1.4).

31.8 cm diameter; United Solutions 5-gallon Residential Bucket, Lowes, Mooresville, NC) each with a polystyrene foam raft (Kingspan Insulation, 1.9 cm × 1.2 m × 2.4 m R-4 Un-face Polystyrene Foam Board Insulation, Winchester, VA) cut to match the size of bucket. Three holes of the size of rockwool cubes (diameter: 2.5 cm) were cut into each raft (three plants per raft). Each DWC unit (16 units in total) contained 15 L of nutrient solution made using dechlorinated and ultraviolet-treated water.

Following Expt. 1, the same nutrient solution recipe was used as the basal formula in this experiment. Additionally, three times higher strength of nutrient solution was examined by adjusting dilution rate of the stock solutions. The resulting EC after mixing with ultraviolet-treated dechlorinated water was either 1.4 or 3.4 dS·m⁻¹. Before transplant, nutrient solution pH was adjusted to setpoints 4.5 or 5.5 using sulfuric acid. At least once a day, pH was monitored and manually adjusted thereafter as needed by adding sulfuric acid to maintain pH within a range of ± 0.25 of target pH. At least three times a week, EC and DO of each DWC unit were measured using the same handheld meters as described before. Nutrient solutions were sent to the same commercial analytical laboratory (JR Peters, Allentown, PA) for micro- and macronutrients at the end of experiment. Twenty days after transplanting, all plants were harvested for quantifying plant growth and nutrient concentrations in the same procedures as in Expt. 1.

This experiment was conducted twice (Trial 1 and 2) each with the four treatments replicated in four blocks inside the growth chamber. Because there were no interactions by trial, plant growth and nutrient data were pooled (n = 8) and subjected to ANOVA followed by *t* test paired with the control (standard EC and pH) and between standard and high EC at low pH. All statistical analyses were performed using JMP software (SAS Institute).

Results and Discussion

Environmental conditions. Average day and night growth chamber air temperatures, nutrient solution temperatures, vapor pressure deficit (VPD), and photosynthetic photon flux density are reported in Table 1. All setpoints were maintained within an acceptable range throughout the experiments. Measured nutrient concentrations in solution at the end of the experiments are reported in Table 2. All macro- and micronutrients were at comparable levels between pH treatments in Expt. 1. In Expt. 2, all macronutrient concentrations except S remained 2.6 to 3.4 times higher in high EC treatments (marked as HE). The S concentration was only 1.9 times greater in high EC treatments likely because of the difference in the amount of sulfuric acid used for pH adjustment. Micronutrient concentrations except Cl and Zn were 2.6 to 4 times higher in high EC treatments than low EC treatments. Cl and Zn concentrations

Table 3. Root-zone environmental parameters recorded in Expts. 1 and 2 (each with two replications). Mean ± SD of daily measurements.

Trial	Treatment code ^z	pH	EC (dS·m ⁻¹)	Dissolved oxygen (ppm)
Expt. 1				
1	pH 4.0	4.03 ± 0.09	1.44 ± 0.06	9.04 ± 0.16
	pH 4.5	4.56 ± 0.11	1.49 ± 0.04	8.88 ± 0.21
	pH 5.0	5.01 ± 0.29	1.49 ± 0.03	8.87 ± 0.21
	pH 5.5	5.44 ± 0.26	1.39 ± 0.04	8.99 ± 0.22
2	pH 4.0	4.04 ± 0.11	1.42 ± 0.04	9.13 ± 0.29
	pH 4.5	4.47 ± 0.17	1.41 ± 0.03	9.03 ± 0.28
	pH 5.0	4.96 ± 0.14	1.41 ± 0.04	9.06 ± 0.43
	pH 5.5	5.47 ± 0.11	1.40 ± 0.00	9.12 ± 0.30
Expt. 2				
1	LP/HE	4.72 ± 0.01	3.43 ± 0.01	7.23 ± 0.32
	LP/SE	4.75 ± 0.02	1.42 ± 0.01	7.06 ± 0.33
	SP/HE	5.54 ± 0.01	3.38 ± 0.01	7.13 ± 0.31
	SP/SE (control)	5.58 ± 0.02	1.40 ± 0.01	7.25 ± 0.33
2	LP/HE	4.63 ± 0.01	3.43 ± 0.01	6.54 ± 0.13
	LP/SE	4.69 ± 0.01	1.39 ± 0.01	6.78 ± 0.22
	SP/HE	5.52 ± 0.06	3.43 ± 0.01	6.50 ± 0.36
	SP/SE (control)	5.54 ± 0.01	1.40 ± 0.00	6.57 ± 0.20

^zTreatments in Expt. 2 are combinations of LP (low pH 4.5), SP (standard pH 5.5), HE (high electrical conductivity 3.4), and SE (standard electrical conductivity 1.4).

were only 1.5 to 2.3 times greater in high EC treatments. Average pH, EC, and DO of experimental nutrient solutions are reported in Table 3 and were in the target ranges of the experiments.

Effects of pH of nutrient solutions on spinach growth and nutrient uptake (Expt. 1). All spinach growth parameters measured in this experiment (shoot and root fresh/dry mass,

and leaf count per plant) declined as lowering pH (Fig. 1). Fresh shoot mass (i.e., yield) of spinach was 0.88 g (12%) and 3.8 g (52%) per plant (averages of two trials) at pH 4.0 and 4.5, respectively, compared with those at the standard pH 5.5 (7.3 g per plant). Similarly, fresh root mass, dry shoot mass, and dry root mass (all per plant) were 0.26 g (11%), 0.12 g (21%), and 0.047 g (33%) at pH 4.0,

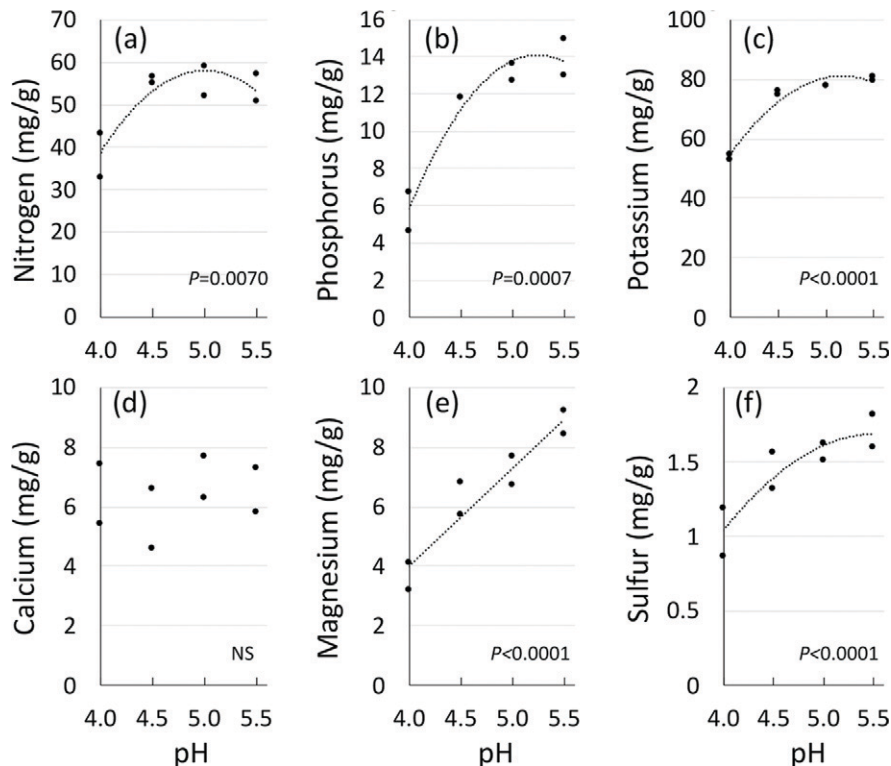


Fig. 2. Responses of leaf nutrient concentrations of spinach plants to nutrient solution pH 4.0, 4.5, 5.0, or 5.5 (Expt. 1). Responses significant by analysis of variance ($P < 0.15$) were expressed with linear or quadratic regressions: (a) $y = -19.3x^2 + 193x - 424$; (b) $y = -5.35x^2 + 56.1x - 132$; (c) $y = -19.2x^2 + 198x - 431$; (d) NS; (e) $y = 3.28x - 9.11$; (f) $y = -0.272x^2 + 3.02x - 6.67$; (g) NS; (h) $y = 0.926x - 2.109$; (i) $y = 18.5x - 9.34$; (j) $y = 22.1x - 48.1$; (k) NS; (l) $y = -94.3x^2 + 1013x - 2481$; (m) $y = -277x + 1692$. Regression P values (NS = nonsignificant at $P < 0.05$) are shown in each chart.

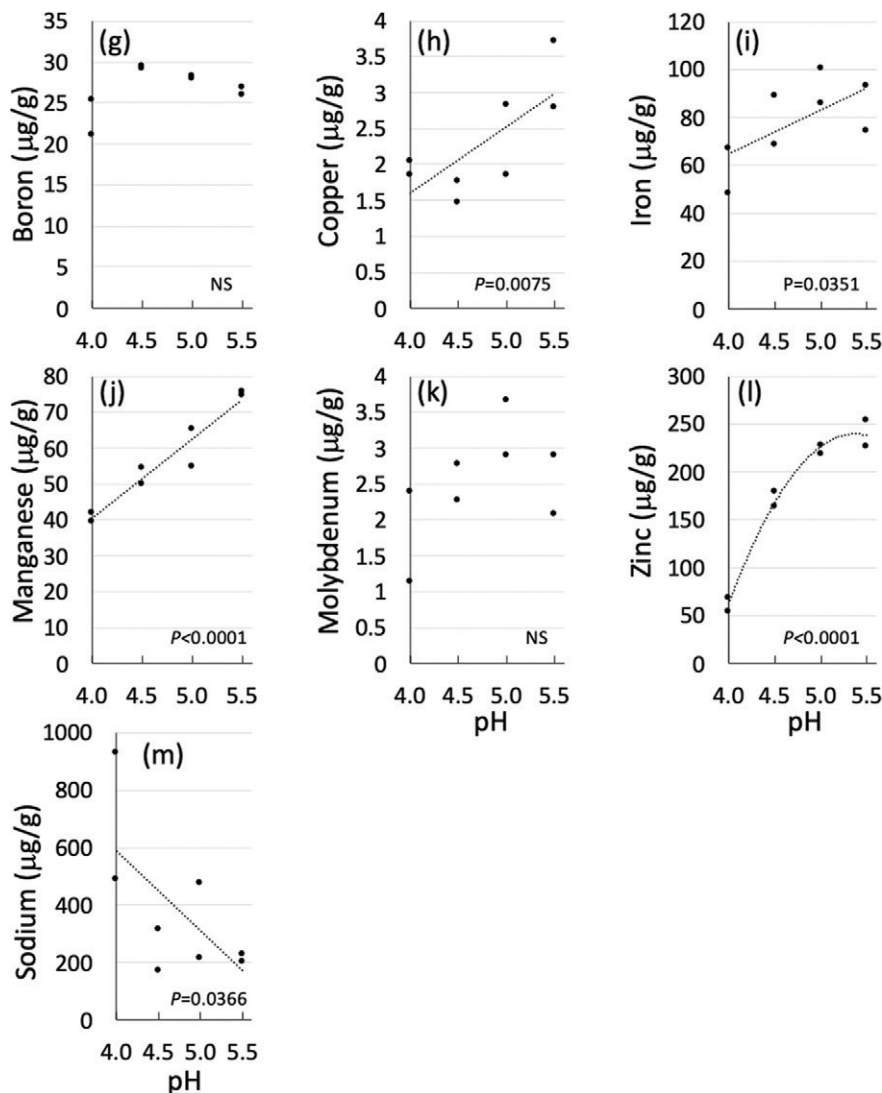


Fig. 2. (Continued).

and 1.6 g (65%), 0.31 g (53%), and 0.11 g (79%) at pH 4.5 of those at the standard pH 5.5 (2.4, 0.58, and 0.14 g), respectively. The reduction of leaf count by lowering pH was significant ($P < 0.0001$) but moderate magnitude ($<25\%$ reduction) than the biomass accumulation ($\approx 70\%$ to 90% reduction). Plant leaf development is strongly affected by temperature (e.g., Walters and Currey, 2019), whereas plant biomass accumulation is determined by photosynthesis and leaf expansion, both of which are affected by plant nutrient status directly. Adversely effects of inadequate pH are well known in hydroponics nutrient management as pH affects availability of many essential nutrients for plant growth (Adams, 2002). For example, Arnon and Johnson (1942) showed that lettuce and tomato plants did not grow at pH 3.0 and exhibited reduced shoot and root fresh weight at pH 4.0 compared with pH 5.0 or 6.0.

In our experiment, spinach plants grown at pH 4.0 showed yellowing of older leaves and were stunted. Their roots were contained within the small rockwool plugs and did not extend into the nutrient solution. In contrast,

plants grown at pH 4.5 extended roots showing normal morphology similar to those at standard pH 5.5. Therefore, we consider that the growth inhibition at pH 4.0 was likely due to the direct damage by high hydronium ion concentrations, whereas that at pH 4.5 was due to the indirect factors such as low nutrient uptake. In our earlier study, plant growth of two cultivars of sweet basil was unaffected by the same range of pH (4.0 to 5.5) (Gillespie et al., 2020). The growth response to pH is species specific and we need further studies to investigate responses to pH of commercially important cultivars and species grown hydroponically. Regardless, this was the first report showing spinach plant growth response to the acidic range of pH.

Dry leaf tissue concentrations of N, P, K, Mg, S, Cu, Fe, Mn, and Zn exhibited a significant decline as lowering pH (Fig. 2). The other elemental concentrations were either not significant (Ca, B, and Mo) or increased (Na). Among these nutrients having the significant reduction by lowering pH, K, Mg, Cu, Fe, Mn, and Zn are taken up by the plants as cations and N (NO_3^-), P, and S are

anions. Cation antagonism can be considered as a main factor affecting reduced uptake of cations under low pH (high hydronium ion concentrations) (Mengel et al., 2001; Peterson et al., 1982). Smith et al. (2004) also observed reduced cation concentrations in geranium plant leaves at low substrate pH and indicated competition between cations and hydrogen ions for root binding sites, low-pH stress on membrane and cation channel activity, or reduced uptake into the shoot tissue. Major macronutrients involved in cation antagonism in hydroponics are NH_4^+ -N, K, Ca, and Mg. In the present experiment, our nutrient solution did not contain NH_4^+ -N (all NO_3^- -N), and all other cation concentrations (except Ca and Na) declined with lowering pH. Unlike other studies reporting recovery of growth by increasing Ca concentration at low pH (e.g., Arnon and Johnson, 1942) or a decline of Ca uptake as lowering pH (e.g., Gillespie et al., 2020), no clear responses of Ca concentration were observed in the present experiment. It is unclear why Ca uptake was unaffected by low pH for spinach but this is possibly due to the preference of Ca among the antagonistic cations for the Ca homeostasis in spinach. Of interest, Ferreira et al. (2020) showed a unique nutrient uptake of spinach plants under combinations of salinity (high Na) and K deficiency, where spinach plants reduced Ca concentration by increased Na but not necessarily by increased K concentration in the nutrient solution. In our experiment, increasing Na as decreasing pH was also observed. According to Ferreira et al. (2020), Na was likely an essential mineral for the growth when K was deficient in the root zone.

Reduction of N, P, and S uptake under low pH may be more associated with overall root function. Tissue concentrations of N, P, and S were relatively similar between pH of 4.5–5.5 and declined at a greater extent at the lowest pH (4.0) where plants were stunted. Therefore, the reduced uptake of these ions might be associated with the direct damage on roots by high hydronium ion concentrations.

In our earlier study (Gillespie, 2019), we had additional treatments examining adjusted micronutrient concentrations; we found significant reduction of B and Mo, and in addition to those, we found significant reduction in the present experiment (N, P, K, Mg, S, Cu, Fe, Mn, and Zn) with lowering pH. Together, the growth reduction we observed in this experiment was likely due to the reduced nutrient uptake by the plants grown under low pH. It should be noted that these overall reductions of nutrient uptake are different from what is reported for soil-based or soilless substrate-based systems. In these systems, typically toxicities of cationic metal ions occur at low pH (e.g., Peterson, 1982; Smith et al., 2004). In liquid-based hydroponics, due to the minimum use of substrates (only 15 to 20 mL volume of rockwool substrate against 4 to 5 L of nutrient solution per plant), interactions with cation exchanges with substrates virtually do not exist. Therefore, effects of pH on nutrient uptake in hydroponics

are different compared with those in soil and soilless substrate culture.

Effects of increased nutrient concentrations on spinach growth and nutrient uptake under low pH (Expt. 2). Spinach plants in all four treatments (pH \times EC) exhibited normal growth without abnormal morphology or symptoms indicating nutrient disorders. However, as seen in Expt. 1, plant growth and development under low pH 4.5 and standard EC 1.4 (marked as LP/SE) was largely reduced compared with those under the control (SP/SE) (Fig. 3). Fresh shoot and root, dry shoot and root mass, leaf count, and leaf area were reduced by 62%, 44%, 55%, 23%, 19%, and 55%, respectively, by lowering pH without altering EC of the nutrient solution. When nutrient concentrations and thereby EC were increased at low pH (LP/HE), these plant growth parameters were significantly increased compared with standard EC at low pH (LP/SE), although they are still significantly lower than those of the control (SP/SE), except the dry root mass ($P = 0.372$). Increasing EC under the standard pH (SP/HE) did not significantly increase plant growth and development compared with the control (SP/SE), suggesting that nutrient levels were sufficient and not limiting the growth of spinach under the standard pH. In contrast, Öztekin et al. (2018) reported that the yield of spinach plants grown in a deep water culture system with a half strength of nutrient solution was 17% lower than that with a full strength solution ($\text{mg}\cdot\text{L}^{-1}$; N 150, P 50, K 150, Ca 150, and Mg 50). Our macronutrient

concentrations of standard EC (Table 2, SE) were lower than what reported as full strength in Öztekin et al. (2018) (except Ca concentration). Furthermore, Öztekin et al. (2018) also showed that the effect of different nutrient concentrations on spinach growth was more pronounced under higher solar radiation and temperature. Therefore, the nonsignificant increase in plant growth by increasing nutrient concentrations (HE) may suggest that nutrient availability and uptake are already saturated at these concentrations under the present conditions. Increasing nutrient concentrations increases the osmotic stresses, which could adversely affect plant growth under high EC. However, the literature provides conflicting results on the salinity tolerance of spinach. For example, 'Crocodile' spinach growth was greatly reduced by irrigation water salinity of $6.5\text{ dS}\cdot\text{m}^{-1}$ (Xu and Mou, 2016), whereas another study (Ferreira et al., 2020) suggests that, using cultivars Raccoon and Gazzelle, a salinity threshold for irrigation water was 7 to $10\text{ dS}\cdot\text{m}^{-1}$ in spinach. Our high EC treatment was $3.4\text{ dS}\cdot\text{m}^{-1}$ and unlikely considered saline to spinach plants.

Under standard EC, low pH of 4.5 (LP/SE) decreased P, K, Ca, Mg, S, Cu, Mn, and Zn concentrations in dry leaf and increased B and Mo concentrations, compared with those in the control (SP/SE) (Table 4). No significant effects of low pH were observed in N, Fe, and Na concentrations under standard EC. Increasing nutrient concentrations at low pH (LP/HE) significantly increased all dry

leaf nutrient concentrations except K, B, Fe, and Zn. Under low pH and increased EC treatment (LP/HE), all dry leaf nutrient concentrations were similar or higher than those of the control except Mg and Zn, which showed a lower concentration than the control with a weak significance ($P = 0.051$ and 0.056 for Mg and Zn, respectively). Calcium concentration exhibited the largest increase by increased EC under low pH (LP/HE), reaching nearly 50% higher concentration of the control plants (SP/SE). This increased Ca uptake may have reduced uptake of antagonistic cations such as Mg by the roots and leaf tissue.

Magnesium is involved in numerous key functions in plants including photosynthesis and loading sucrose to phloem (Cakmak and Yazici, 2010; Guo et al., 2016). Leaf Mg concentrations of spinach plants grown in greenhouse were reportedly at ≈ 8 to $10\text{ mg}\cdot\text{g}^{-1}$ without salinity (Ferreira et al., 2020). We did not notice a typical symptom of Mg deficiency in this experiment (e.g., interveinal yellowing of lower leaves). However, Mg deficiency-caused abnormal physiology may exist before a typical visual symptom appears. For example, Hermans et al. (2004) showed a large accumulation of sucrose in Mg-deficient plant leaves of *Beta vulgaris* before any loss in photosynthetic activity or reduction of biomass. Critical leaf Mg thresholds that cause yield reduction are species specific. Hauer-Jákli and Tränkner (2019) reported Mg thresholds of various crops species. Although spinach was not included in their report, the thresholds were shown in the range of 1 to $2\text{ mg}\cdot\text{g}^{-1}$, which is lower than the Mg concentrations observed in our experiment. Therefore, it is unlikely that our plants under LP/HE treatment are considered as exhibiting Mg deficiency. Nevertheless, further possible improvement of spinach plant growth at varied concentrations of Mg in the low pH nutrient solution is necessary.

Typical Zn deficiency reported for other species includes stunted growth with small leaves, which may be the result from loss of the capacity to produce significant amounts of auxin indole-3-acetic acid (Taiz and Zeiger, 2006). However, the spinach plants in this experiment did not show stunted growth in any of four treatments, and so it is unlikely that relatively low Zn is the main limiting factor of the plant growth. Moreover, Zn concentration of LP/HE treatment was similar level as that in SP/HE treatment, where plant growth was not significantly different from the control (SP/SE) (Table 4). Therefore, it is unlikely that the Zn was limiting the spinach growth under increased nutrient concentrations (LP/HE). However further optimization of nutrient formula with independently increased nutrients will help better understand the nutrient limiting spinach growth and means to recover the uptake and thereby overall growth of spinach plants. As nutrient uptake, especially Ca and Mg, in leaf tissue is also enhanced by mass flow driven by transpiration, further investigation should consider environmental conditions such as high light, temperature, and

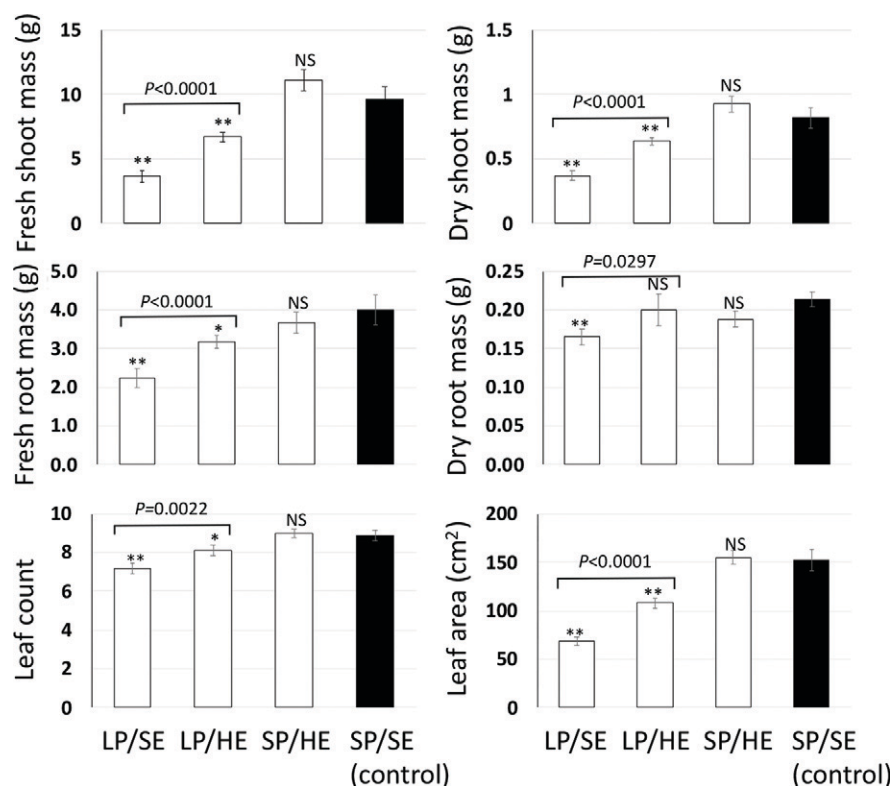


Fig. 3. Spinach plant growth as affected by high electrical conductivity (EC) when grown at low or standard pH of nutrient solution (Expt. 2). Means and SES ($n = 8$). Pairwise comparisons by t test with the control treatment (standard pH and standard EC) are shown (NS = nonsignificantly different at $P = 0.05$; *, ** = significantly different at $P < 0.01$ or 0.05 , respectively). P values of additional pairwise comparisons between standard and high EC at low pH (LP/SE vs. LP/HE) are shown.

Table 4. Spinach leaf nutrient concentrations as affected by high electrical conductivity (EC) and low pH compared with the control (standard pH and standard EC) (Expt. 2). Pairwise comparisons by *t* test with the control treatment (standard pH and standard EC) are shown (NS = nonsignificantly different at $P = 0.05$; *, ** = significantly different at $P < 0.01$ or 0.05 , respectively). *P* values of additional pairwise comparisons between standard and high EC at low pH (LP/SE vs. LP/HE) are shown.

Treat. code ^z	Macronutrients (mg g ⁻¹)							Micronutrients (μg g ⁻¹)						
	N	P	K	Ca	Mg	S	B	Cu	Fe	Mn	Mo	Zn	Na	
LP/SE	54.2 ± 0.39 ^{NS}	12.3 ± 0.58 ^{**}	104.8 ± 8.71 ^{**}	7.25 ± 0.32 [*]	5.93 ± 0.58 ^{**}	4.48 ± 0.15 [*]	46.1 ± 3.34 ^{**}	4.74 ± 0.26 ^{**}	178.8 ± 19.69 ^{NS}	45.5 ± 2.41 ^{**}	8.37 ± 0.75 ^{**}	143.7 ± 6.50 ^{**}	478.5 ± 35.01 ^{NS}	
LP/HE	55.6 ± 0.45 ^{NS}	17.1 ± 0.20 ^{NS}	122.7 ± 5.01 ^{NS}	12.38 ± 0.88 ^{**}	8.16 ± 0.84 ^{NS}	5.62 ± 0.09 ^{NS}	44.2 ± 4.1 [*]	8.31 ± 0.44 ^{NS}	222.5 ± 19.28 ^{NS}	84.0 ± 6.60 ^{NS}	7.50 ± 0.61 [*]	197.2 ± 13.27 ^{NS}	450.7 ± 26.87 ^{NS}	
SP/HE	57.0 ± 0.28 ^{**}	15.2 ± 0.95 ^{NS}	128.3 ± 6.02 ^{NS}	10.12 ± 0.28 ^{**}	11.45 ± 0.77 ^{NS}	5.05 ± 0.13 ^{NS}	36.4 ± 2.88 ^{NS}	9.34 ± 0.22 [*]	204.8 ± 26.41 ^{NS}	124.5 ± 4.34 ^{**}	4.74 ± 0.22 ^{NS}	202.4 ± 12.73 ^{NS}	401.7 ± 13.62 ^{NS}	
SP/SE	54.9 ± 0.45	15.4 ± 0.84	125.3 ± 6.96	8.31 ± 0.46	10.49 ± 0.70	5.07 ± 0.11	32.4 ± 2.88	7.04 ± 0.14	193.0 ± 32.01	97.3 ± 5.17	5.46 ± 0.55	244.4 ± 18.39	446.9 ± 22.91	
(Control)														
<i>P</i> (LP/SE vs. LP/HE)	0.038	0.003	0.089	<0.001	0.046	0.004	0.723	0.001	0.135	<0.001	0.383	0.003	0.539	

^zTreatments in Expt. 2 are combinations of LP (low pH 4.5), SP (standard pH 5.5), HE (high EC 3.4), and SE (standard EC 1.4).

VPD. Foliar nutrient applications are another approach shown as alternative means to increase Mg concentrations in spinach leaves when enhancing uptake by roots is a challenge (Borowski and Michalek, 2010).

Increasing EC under standard pH 5.5 (SP/HE) increased N, Ca, Cu, and Mn concentrations but did not affect the other nutrient concentrations (Table 4). As described earlier, these increases in N, Ca, Cu, and Mn did not reflect plant growth as nutrient concentrations were generally sufficient in the standard pH conditions regardless of EC.

Conclusion

Spinach plant growth was significantly decreased by lowering nutrient solution pH. Elemental concentrations of N, P, K, Mg, S, Cu, Fe, Mn, and Zn in plant leaves were lowest when grown in pH 4.0. Stunted growth was only observed in spinach grown in pH 4.0, likely because of the direct effect of hydronium ion damage. Spinach grown in pH 4.5, 5.0, and 5.5 all displayed normal shoot and root growth, suggesting that reductions in shoot growth at pH 4.5 and 5.0 were attributed to decreased nutrient uptake as opposed to the direct effect of pH. Increased nutrient concentrations (EC 3.4 dS·m⁻¹) effectively increased the plant growth under low pH 4.5 but did not fully recover the shoot fresh and dry weight compared with those of control plants (standard EC 1.4 dS·m⁻¹ and standard pH 5.5). Further optimization of individual nutrient concentrations need to be conducted to better understand the dynamics of nutrient uptake under low pH, which may allow spinach to be grown without significant reductions in shoot growth in low pH as potential means of low-cost disease control.

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